<u>LETTERS</u>

Total Synthesis and Determination of the Absolute Configuration of Vinylamycin

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Supporting Information

ABSTRACT: The absolute configurations of the three unknown chiral centers in vinylamycin were predicted according to the structural comparison with microtermolide A and rakicidin A, and then total syntheses of vinylamycin were applied to determine the three unknown chiral centers as 14*R*, 15*R*, and 16*S*.

T he antimicrobial natural product, vinylamycin, was discovered as a metabolite of *Streptomyces sp.* and was initially proposed to be a 14-membered ring with five chiral centers (Scheme 1).¹ Vinylamycin belongs to a family of cyclic

Scheme 1. Structure of Cyclic Despipeptides Natural Products



depsipeptides, which possess a unique 4-amino-2,4-pentadienolate structure, and include rakidicins,² microtermolide A,³ and BE43547A1.⁴ Using degradation and chiral HPLC analysis, the amino acids of vinylamycin at C-2 and C-5 were identified as D-valine (2*R*) and L-alanine (5*S*);¹ however, its three consecutive chiral centers, C-14, C-15, and C-16 in the polyketide fragment, have remained unknown. In 2012, another similar natural product, microtermolide A, was isolated and extensively studied with NMR.³ As a result, the three absolute configurations of microtermolide A were tentatively assigned as 14*R*, 15*R*, and 16*R*,³ thereby representing an *anti–anti* relative



relationship among the chiral centers. Based on this analysis of microtermolide A and prediction with calculation, the structure of vinylamycin was revised to a 15-member ring,³ and then the absolute configuration of vinylamycin could also be tentatively assigned as 14*R*, 15*R*, and 16*R* (compound 1).

Recently, *anti*-cancer stem cell natural product rakicidin A was prepared through total synthesis in our group. The absolute configuration of rakicidin A was determined to be 2*S*, 3*S*, 14*S*, 15*S*, and 16*R*; thus, the three consecutive chiral centers exhibited an *anti*-*syn* relative relationship.^{5a} This result was further approved by degradation and HPLC analysis.^{5b} Furthermore, many other depsipeptides with three consecutive chiral centers and long lipophilic side chains, including emericellaminde A,⁶ stevastelin C3,⁷ and arthrichitin,⁸ have an *anti*-*syn* polyketide carbon framework. Therefore, it was also possible that vinylamycin exhibited *anti*-*syn* relative stereo-chemistry, i.e., 14*R*, 15*R*, and 16*S* for its three consecutive chiral centers (compound 2).

In contrast to the total synthesis of rakicidin A,^{5a} the synthetic strategy of vinylamycin was modified such that lactamization was performed at C-7, which involved a less hindered carboxylic acid. Compound 4a/4b could be prepared from the aldol product 5a/5b, which may be obtained from the L-camphorsultam derivative 10 and the chiral aldehyde 9a/9b. Due to the liability of the conjugated diene group, it was necessary to produce it in the final steps of the synthesis. Fragment 7 was prepared in prior steps using D-serine to overcome the low yield and low selectivity of the HWE reaction in this complex system^{5a} (Scheme 2).

The compound **15** with the desired configuration was synthesized with high diastereoselectivity via a Mukaiyama aldol reaction, and the configuration was confirmed by X-ray analysis

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Scheme 2. Retrosynthetic Analysis



of the single crystal of 15 (refer to the Supporting Information). The high dr may be explained by the fact that the process underwent the proposed transition state 14. Thus, the desired products 5a/5b were synthesized with good selectivity (dr > 10:1) and moderate yields following the same route (Scheme 3). Hydrolysis of 5a/5b followed by treatment with Cs_2CO_3 and allyl bromide achieved the allyl ester 17a/17b.

Scheme 3. Aldol Reaction To Form Polyketide Fragments



For the synthesis of the required serine derivative 7, aldehyde 18^{12} was subjected to the highly selective HWE reaction (*E*:*Z* > 25:1) to provide the conjugated alkene 20 (Scheme 4).¹³ Deprotection of both Boc and TBS groups in compound 20 was followed by reprotection of the OH group with TBS to obtain compound 7.¹⁴



Using compounds 17a/17b and 7, an efficient synthetic approach for fragment coupling was investigated (Scheme 5). Esterification of alcohol 17a/17b with the D-Fmoc-valine 8 produced compounds 21a and 21b. Deprotection of the Fmoc group and coupling with L-Fmoc-alanine 6 provided compounds 23a/23b. Thus, deprotection of the allyl groups,





followed by coupling with the D-serine derivative 7, resulted in the cyclization precursors, 4a/4b. Following cleavage of the allyl ester and Fmoc protecting group, the resulting intermediates were used directly in the next lactamization with the previously optimized conditions to produce lactam 3a and 3b in 49% and 45% yield, respectively.^{5a} Exposure of cyclic compounds 3a and 3b to HF provided free alcohols 24a/24b. With subsequent mesylation, the intermediates 25a/25b were achieved. Deprotection of PMB with DDQ and final elimination of the Ms group with DBU furnished compounds 2 and 1, respectively. Of note, both compounds were not very stable at room temperature.

NMR analytic data of the compounds 1 and 2 were compared with those of natural vinylamycin reported in the literature,¹ including both the ¹H and ¹³C NMR data (Supporting Information). The ¹H NMR analytic data of reported vinylamycin, compound 1, and compound 2 were very similar, suggesting that prediction of stereochemistry outside the macrocyclic ring using calculation was difficult. However, the data for compound 2 and natural vinylamycin matched perfectly, while the data for compound 1 clearly differed at C-15, C-16, C-17, C-20, C-22, and C-23 in the ¹³C NMR. These results indicated that the structure of compound 2 represented the real structure of natural vinylamycin. Consequently, the three previously unknown consecutive chiral centers of vinylamycin were assigned as 14*R*, 15*R*, and 16*S*, thereby representing *anti–syn* relative stereochemistry.

In conclusion, the structure of the natural cyclic depsipeptide, vinylamycin, was recently revised to a 15-member ring. However, three consecutive chiral centers in vinylamycin remained unknown,^{1,3} and vinylamycin possibly represented one of eight possible diastereoisomers. Here, a structural comparison of microtermolide A³ and rakicidin A⁵ with vinylamycin performed respectively to predict the stereochemistry of three consecutive chiral centers. The difference between the two predictions is an anti-anti or anti-syn relative stereochemistry in the three consecutive chiral centers, respectively. Therefore, both predicted compound 1 (antianti) and compound 2 (anti-syn) underwent total syntheses. Both compounds 1 and 2 were synthesized in a highly efficient manner in 16 linear steps with a 3.7% overall yield, which is significantly higher than the overall yield (0.17%) of total synthesis of rakicidin A.5a When the analytic data for

Organic Letters

compounds 1 and 2 were compared with those for natural vinylamycin,¹ only the analytic data for compound 2 perfectly matched those reported for vinylamycin. Therefore, the remaining three unknown consecutive chiral centers of vinylamycin should be assigned as 14R, 15R, and 16S, thereby identifying vinylamycin as another example of *anti*—*syn* relative stereochemistry in a cyclic depsipeptide with three consecutive chiral centers and long lipophilic side chains. The determination of three unknown consecutive chiral centers and the highly efficient synthetic sequence of vinylamycin provide the opportunity for further investigation of vinylamycin and similar depsipeptides as interesting biologically active natural products.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.5b02809.

Experimental details and characterization of all new compounds, including ¹H, ¹³C, and selected X-ray crystal structure, 2D-HMQC, and HMBC data (PDF)

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Igarashi, M.; Shida, T.; Sasaki, Y.; Kinoshita, N.; Naganawa, H.; Hamada, M.; Takeuchi, T. J. Antibiot. **1999**, *52*, 873–879.

(2) (a) McBrien, K. D.; Berry, R. L.; Lowe, S. E.; Neddermann, K. M.; Bursuker, I.; Huang, S.; Klohr, S. E.; Leet, J. E. J. Antibiot. 1995, 48, 1446–1452. (b) Hu, J. F.; Wunderlich, D.; Sattler, I.; Feng, X. Z.; Grabley, S.; Thiericke, R. Eur. J. Org. Chem. 2000, 2000, 3353–3356. (c) Igarashi, Y.; Shimasaki, R.; Miyanaga, S.; Oku, N.; Onaka, H.; Sakurai, H.; Saiki, I.; Kitani, S.; Nihira, T.; Wimonsiravude, W.; Panbangred, W. J. Antibiot. 2010, 63, 563–565.

(3) Carr, G.; Poulsen, M.; Klassen, J. L.; Hou, Y. P.; Wyche, T. P.; Bugni, T. S.; Currie, C. R.; Clardy, J. Org. Lett. 2012, 14, 2822–2825.
(4) Nishioka, H.; Nakajima, S.; Nagashima, M.; Ojiri, K.; Suda, H. IP10147594-A, 1998.

(5) (a) Sang, F.; Li, D. M.; Sun, X. L.; Cao, X. Q.; Wang, L.; Sun, J. L.; Sun, B. X.; Wu, L. L.; Yang, G.; Chu, X. Q.; Wang, J. H.; Dong, C. M.; Geng, Y.; Jiang, H.; Long, H. B.; Chen, S. J.; Wang, G. Y.; Zhang, S. Z.; Zhang, Q.; Chen, Y. J. Am. Chem. Soc. 2014, 136, 15787–15791.
(b) Oku, N.; Matoba, S.; Yamazaki, Y. M.; Shimasaki, R.; Miyanaga, S.; Igarashi, Y. J. Nat. Prod. 2014, 77, 2561–2565.

(6) Oh, D. C.; Kauffman, C. A.; Jensen, P. R.; Fenical, W. J. Nat. Prod. 2007, 70, 515–520.

(7) Kurosawa, K.; Matsuura, K.; Nagase, T.; Chida, N. Bull. Chem. Soc. Jpn. 2006, 79, 921–937.

(8) Schlingmann, G.; Milne, L.; Williams, D. R.; Carter, G. T. J. Antibiot. 1998, 51, 303-316.

(9) Prantz, K.; Mulzer, J. Chem. - Eur. J. 2010, 16, 485-506.

(10) Sedrani, R.; Kallen, J.; Cabrejas, L. M. M.; Papageorgiou, C. D.; Senia, F.; Rohrbach, S.; Wagner, D.; Thai, B.; Eme, A. M. J.; France, J.; Oberer, L.; Rihs, G.; Zenke, G.; Wagner, J. J. Am. Chem. Soc. 2003, 125, 3849–3859.

(11) Poulsen, T. B. Chem. Commun. 2011, 47, 12837-12839.

(12) Cook, G. R.; Shanker, P. S. J. Org. Chem. 2001, 66, 6818–6822.
(13) Wadsworth, J. M.; Clarke, D. J.; McMahon, S. A.; Lowther, J. P.; Beattie, A. E.; Langridge-Smith, P. R. R.; Broughton, H. B.; Dunn, T. M.; Naismith, J. H.; Campopiano, D. J. J. Am. Chem. Soc. 2013, 135, 14276–14285.

(14) Palomo, C.; Aizpurua, J. M.; Balentova, E.; Jimenez, A.; Oyarbide, J.; Fratila, R. M.; Miranda, J. I. Org. Lett. 2007, 9, 101–104.

(15) Takeuchi, M.; Ashihara, E.; Yamazaki, Y.; Kimura, S.; Nakagawa, Y.; Tanaka, R.; Yao, H.; Nagao, R.; Hayashi, Y.; Hirai, H.; Maekawa, T. *Cancer Sci.* **2011**, *102*, 591–596.